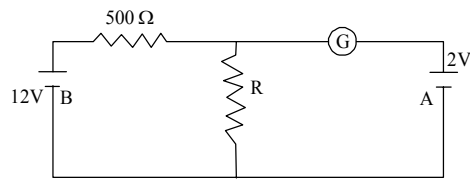
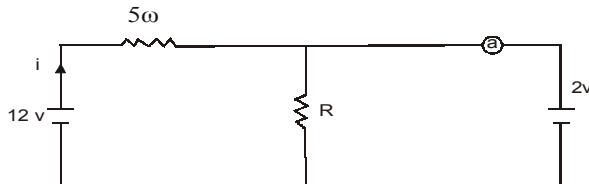


61. In the circuit, the galvanometer G shows zero deflection. If the batteries A and B have negligible internal resistance, the value of the resistor R will be -



- 1) 100  $\Omega$                       2) 200  $\Omega$                       3) 1000  $\Omega$                       4) 500  $\Omega$

**Sol:** 1)



$$12 - 2 = (5\omega)i$$

$$i = \frac{1}{50}$$

$$I = \frac{12}{5\omega} = \frac{1}{5_0} = \frac{12}{5\omega + R}$$

$$5\omega + R = 6\omega = R = 100 \Omega$$

62. A moving coil galvanometer has 150 equal divisions. Its current sensitivity is 10 divisions per milliampere and voltage sensitivity is 2 divisions per millivolt. In order that each division reads 1 volt, the resistance in ohms needed to be connected in series with the coil will be -

- 1)  $10^5$                       2)  $10^3$                       3) 9995                      4) 99995

**Sol:** 3)

$$\text{Voltage sensitivity} = \frac{\text{Current sensitivity}}{\text{Resistance of galvanometer } G} \Rightarrow G = \frac{10}{2} = 5\Omega$$

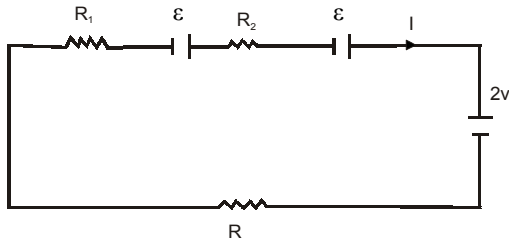
$$\text{Here } i_g = \text{Full scale deflection current} = \frac{150}{10} = 15mA$$

$$V = \text{voltage to be measured} = \frac{150}{15 \times 10^{-3}} - 5 = 9995\Omega$$

63. Two sources of equal emf are connected to an external resistance R. The internal resistances of the two sources are  $R_1$  and  $R_2$  ( $R_2 > R_1$ ). If the potential difference across the source having internal resistance  $R_2$  is zero, then

- 1)  $R = R_2 - R_1$                       2)  $R = R_2 \times (R_1 + R_2) / (R_2 - R_1)$   
 3)  $R = R_1 R_2 / (R_2 - R_1)$                       4)  $R = R_1 R_2 / (R_1 + R_2)$

Sol: 1)



$$I = \frac{2\varepsilon}{R + R_1 + R_2}$$

Potential difference across (L) cell

$$= V = \varepsilon - i R_2 = 0$$

$$\varepsilon - \frac{2\varepsilon}{R + R_1 + R_2} \cdot R_2 = 0$$

$$R + R_1 + R_2 - 2R_2 = 0$$

$$R + R_1 - R_2$$

$$\therefore R = R_2 - R_1$$

64. Two voltmeters, one of copper and another of silver, are joined in parallel. When a total charge  $q$  flows through the voltmeters, equal amount of metals are deposited. If the electrochemical equivalents of copper and silver are  $z_1$  and  $z_2$  respectively the charge which flows through the silver voltmeter is

- 1)  $\frac{q}{1 + \frac{z_2}{z_1}}$       2)  $\frac{q}{1 + \frac{z_1}{z_2}}$       3)  $q \frac{z_2}{z_1}$       4)  $q \frac{z_1}{z_2}$

Sol: 1)

$$m = zq \Rightarrow z \propto \frac{1}{q} \Rightarrow \frac{z_1}{z_2} = \frac{q_2}{q_1} \dots (i)$$

$$\text{also } q = q_1 + q_2 \Rightarrow \frac{q}{q_2} = \frac{q_1}{q_2} + 1 \Rightarrow q_2 = \frac{q}{1 + \frac{q_1}{q_2}} \dots (ii)$$

From equation (1) and (ii)  $q_2 = \frac{q}{1 + \frac{z_2}{z_1}}$

65. Two concentric coils each of radius equal to  $2\pi$  cm are placed at right angles to each other. 3 ampere and 4 ampere are the currents flowing in each coil respectively. The magnetic induction in Weber/m<sup>2</sup> at the centre of the coils will be

$$(\mu_0 = 4\pi \times 10^{-7} \text{ Wb / A.m.})$$

- 1)  $10^{-5}$       2)  $12 \times 10^{-5}$       3)  $7 \times 10^{-5}$       4)  $5 \times 10^{-5}$





75. A coil of inductance 300 mH and resistance  $2\Omega$  is connected to a source of voltage 2 V. The current reaches half of its steady state value in

1) 0.1 s

2) 0.05 s

3) 0.3 s

4) 0.15 s

**Sol:** 1)

$$i = i_0 \left( 1 - e^{-\frac{t}{L}} \right)$$

$$\frac{i_0}{2} = i_0 (1 - e^{-t})$$

$$\frac{L}{R} \times 0.693 = t$$

$$= 0.693 \times \frac{300 \times 10^{-3}}{2}$$

$$= 0.1 \text{ sec}$$

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